



Calculation the Nuclear Radiation Absorption Parameters in Pb Alloys

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Abstract: In this research, various binary alloys were prepared and studied, consisting of lead as the matrix material, with different weight percentages of bismuth and tin added as enhancing elements. The study focused on evaluating the physical properties of these alloys, specifically their ability to attenuate gamma rays. For this purpose, the mass attenuation coefficient (μ_m) of the prepared alloys was calculated using the Nist-XCom database, over a wide energy range from 0.05953 to 800(MeV). Based on the calculated mass attenuation coefficient values, a set of other important attenuation parameters were derived, including the linear attenuation coefficient (μ), molecular cross section (σ_m), electron cross section (σ_{ele}), effective atomic number (Z_{eff}), electron density (N_{ele}), mean free path (λ), half-layer thickness (HVL), and tenth-layer thickness (TVL). In addition, the K-edge absorption edges of the elements in the studied alloys were determined.

Keywords: Gamma ray, Half, tenth, quarter, layer, Linear attenuation coefficient, Mass attenuation coefficient, Molecular cross-section, NIST-XCOM.

1. Introduction

To ensure the safe use of ionizing radiation, researchers are increasingly focusing on developing effective shielding materials. The effectiveness of these materials is measured by their ability to reduce radiation transmittance. Therefore, extensive studies are being conducted on the radiation attenuation properties of various materials to identify the most efficient ones. The main physical properties evaluated include the mass and linear attenuation coefficient, molecular and electron interaction cross-sectional area, effective atomic number, electron density, mean free path, as well as half- and decimal layer thicknesses [1].

In 2008, Saxby-Brown et al. conducted a study to investigating the correlation between linear attenuation coefficients and Hounsfield units (HUs). Six different materials were examined, representing the range of values found in clinical practice [2].

A study conducted by Ali H. Taqi et al. in 2021 on ternary alloys of (lead, tin, copper) at varying weight ratios determined their protective properties. These properties included molecular and electronic cross sections (σ_m and σ_{ele}), which were consistent with known experimental values [3].

Using the 662 keV gamma ray backscatter technique, Sharma et al. in 2016 were able to calculate the atomic numbers of a group of binary alloys. The study included lead-tin, lead-zinc, and tin-zinc alloys with varying weight ratios [4].

According to a 2016 study, Kaur et al. calculated the average free path values for several types of heavy metal oxide glasses [5].

In a study published in 2009, Han and Demir were able to determine the values of two critical parameters in metal alloys, namely the effective atomic number and electron density, in alloys consisting of titanium (Ti), copper (Cu), and nickel (Ni) [6].

Tejbir Singh and his research team in 2018 prepared binary lead-copper alloys of different weight ratios, analyzed their physical properties and gamma ray attenuation parameters, especially half-thickness and ten-thickness layers [7].

2. Materials and Methods

In the current study, calculations were performed for some binary metal alloys with varying proportions of their components to form mixtures to select the best of these materials for the manufacture of gamma ray shields. The atomic numbers of the selected elements in the alloy composition are: Pb=82, Sn=50 and Bi=83.

The density of each alloy was calculated from the law for finding the density of mixtures and according to the following equation :

$$P_w = \sum \rho_i * w_i = \rho_1 w_1 + \rho_2 w_2 + \rho_3 w_3 + \dots \quad (1)$$

The molar mass of each alloy can be calculated from the law for finding the molar mass of mixtures and according to the following equation :

$$M_w = \sum M_i * w_i = M_1 w_1 + M_2 w_2 + M_3 w_3 + \dots \quad (2)$$

Table 1: Molecular weights and densities of the alloys under study.

Binary alloys	Samples	%	$\rho \left(\frac{g}{cm^3} \right)$	M (g/mole)
Pb-Sn	S1	60Pb40Sn	9.728	171.804
	S2	70Pb30Sn	10.131	180.653
	S3	80Pb20Sn	10.534	189.502
Pb-Bi	S4	60Pb40Bi	10.716	207.912
	S5	70Pb30Bi	10.872	207.734
	S6	80Pb20Bi	11.028	207.556

The results showed a direct relationship between the lead concentration in the alloys and both the density and theoretical molecular weight, with both properties gradually increasing with increasing lead content. This behavior is explained by the physical nature of lead, which has a higher density and atomic mass than the other alloy components, which positively impacts the overall values of these parameters as its content increases.

The mass attenuation coefficient (μ_m) is a measure used to determine the probability of an incident photon interacting with a material, and is measured in cm^2/g . This coefficient is a non-constant quantity, as it varies depending on the energy of the incident photon and the atomic number of the elements or the effective atomic number in the case of compounds and mixtures, while it is not affected by the physical state of the material. It is calculated using the NIST-XCom database for the elements ^{82}Pb , ^{83}Bi , and ^{50}Sn , over an energy range of 0.05953 to 800 MeV [10].

Molecular cross-sectional area indicates the extent to which the molecules of a substance are able to interact with the incoming particle, and is used as an indicator of the likelihood of this interaction occurring [11].

$$\sigma_m = \mu_m \frac{M}{N_A} \quad (3)$$

The electron cross-sectional area indicates the extent to which an interaction between an incident particle and the orbital electrons of an atom of the target material is likely to occur [11].

$$\sigma_{ele} = \frac{1}{N_A} \sum_i f_i \frac{A_i}{Z_i} (\mu_m)_i \quad (4)$$

In mixtures the sum of the number of atoms is always one $\sum_i n_i = 1$, so the values of σ_m and σ_a are equal

$$\sigma_m = \sigma_a [12].$$

Linear attenuation coefficient (μ) The probability of a photon loss of a gamma ray beam is a result of various reactions as it passes through the material, under ideal geometric conditions for the measurement process, as it is assumed that any reaction can contribute to removing the photon from the radiation beam[11].

$$\mu = \mu_m \chi \rho \quad (5)$$

The atomic number (Z_{eff}) represents a value close to the true atomic number of elements, and is used as a main indicator to assess the ability of materials to protect against radiation. It is also an important tool for analyzing gamma rays when it penetrates into composite materials or mixtures[12].

$$Z_{eff} = \frac{\sigma_a}{\sigma_e} \quad (6)$$

The electron density is an important factor in radiation protection, as it represents the average number of electrons per unit mass in the reaction subject[12].

$$N_{ele} = \frac{\mu_m}{\sigma_{ele}} \quad (7)$$

The mean free path (MFP) expresses the expected distance a photon travels within a material before undergoing an interaction such as scattering or absorption, and is usually measured in centimeters [13].

$$\lambda = \frac{1}{\mu} \quad (8)$$

Half-value layer (HVL) [14]

$$HVL = \ln(2)/\mu \quad (9)$$

Tenth-value layer (TVL)

$$TVL = \ln(10)/\mu \quad (10)$$

3. Calculation

Mass attenuation coefficient values were calculated from the Nist XCom database for ^{82}Pb , ^{83}Bi , and ^{50}Sn in the energy range of (0.05953-800) MeV.

Table 2: Mass attenuation coefficient values were calculated using the Nist XCom database .

E(MeV)	μ (cm ² /g)					
	S1	S2	S3	S4	S5	S6
0.05953	5.755	5.597	5.439	5.209	5.187	5.166
0.08099	2.58	2.522	2.463	2.386	2.376	2.366
0.12178	2.433	2.67	2.907	3.428	3.416	3.405
0.24469	0.4655	0.5056	0.5457	0.6346	0.6324	0.6303
0.35601	0.225	0.2406	0.2562	0.2909	0.29	0.2891
0.364	0.2166	0.2314	0.2461	0.279	0.2781	0.2773
0.66166	0.09636	0.09982	0.1033	0.1112	0.1109	0.1107
0.7789	0.08189	0.08424	0.08659	0.092	0.09182	0.09165
0.96413	0.06782	0.06923	0.07065	0.07396	0.07384	0.07372

1.17323	0.05815	0.05905	0.05995	0.06208	0.062	0.06191
1.25	0.05563	0.05641	0.05719	0.05907	0.05899	0.05891
1.3325	0.05338	0.05407	0.05476	0.05643	0.05636	0.05629
1.40801	0.05165	0.05229	0.05292	0.05445	0.05439	0.05432
2.75	0.0407	0.04123	0.04176	0.04299	0.04295	0.0429
7.12	0.0419	0.04279	0.04368	0.04566	0.04561	0.04556
10	0.04541	0.04649	0.04757	0.04993	0.04988	0.04983
20	0.05588	0.05742	0.05897	0.06234	0.06227	0.0622
30	0.06294	0.06476	0.06658	0.07055	0.07047	0.07039
40	0.06804	0.07006	0.07207	0.07646	0.07637	0.07628
50	0.07192	0.07408	0.07624	0.08094	0.08085	0.08075
60	0.07502	0.07728	0.07955	0.08448	0.08438	0.08428
70	0.07753	0.07989	0.08224	0.08737	0.08726	0.08716
80	0.07963	0.08206	0.08449	0.08977	0.08966	0.08955
90	0.08142	0.08391	0.08639	0.0918	0.09169	0.09158
100	0.08296	0.0855	0.08803	0.09354	0.09343	0.09332
200	0.09148	0.09428	0.09709	0.1032	0.1031	0.1029
300	0.09525	0.09816	0.1011	0.1074	0.1073	0.1072
400	0.09742	0.1004	0.1034	0.1099	0.1097	0.1096
500	0.09885	0.1019	0.1049	0.1115	0.1113	0.1112
600	0.09989	0.1029	0.106	0.1126	0.1125	0.1123
700	0.1007	0.1037	0.1068	0.1135	0.1133	0.1132
800	0.1013	0.1044	0.1074	0.1142	0.114	0.1139

Fig 1 shows the absorption edges (K-edge), known as the Compton edges, for the studied samples. These edges represent the sharp dip in the gamma-ray absorption spectrum resulting from the interaction of photons with the energy levels of electrons within atoms. This occurs when the energy of the incident photon approaches the binding energy of the electrons in the atomic shell. This sudden change in attenuation is attributed to photoelectric absorption, which requires the energy of the photons to be slightly higher than the binding energy of the electrons [15].

The absorption edges of lead metal are clearly visible, with the M edge located between 0.002484 and 0.003851, the L edge between 0.01304 and 0.01586, and the K edge located at 0.088. In bismuth, the M edge lies between 0.00258 and 0.003999, the L edge between 0.01342 and 0.01639, and the K edge is recorded at 0.09053. Tin, on the other hand, has an L edge between 0.003929 and 0.004465, and a K edge at 0.2920.

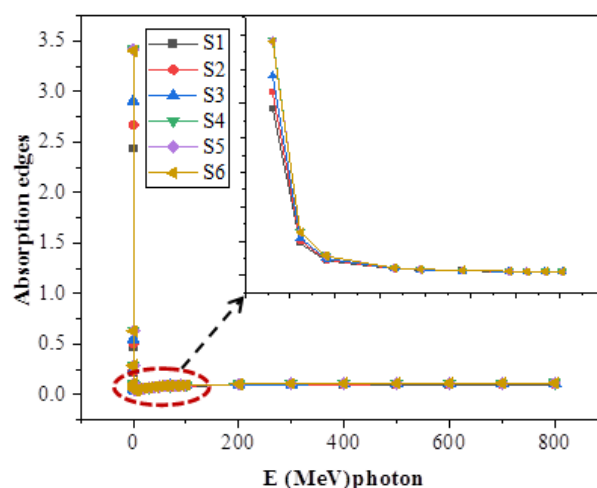


Fig 1. Absorption edges of the samples under study.

Fig 2 illustrates the behavior of the mass attenuation coefficient of the studied samples depending on the incident photon energy. The graph shows an inverse relationship between the mass attenuation coefficient and photon energy; the higher the incident photon energy, the lower the attenuation, and vice versa.

The results also showed that sample S4, composed of (60Pb40Bi), recorded the highest mass attenuation coefficient value. This is attributed to its high content of lead and bismuth, known for their high density, large atomic numbers, and large interaction cross-sectional areas, which enhance the material's ability to absorb gamma rays. In contrast, sample S1, composed of (60Pb40Sn), recorded the lowest mass attenuation coefficient value due to its lower lead content compared to the other samples and its higher tin content, which has a lower atomic number and lower density than lead and bismuth.

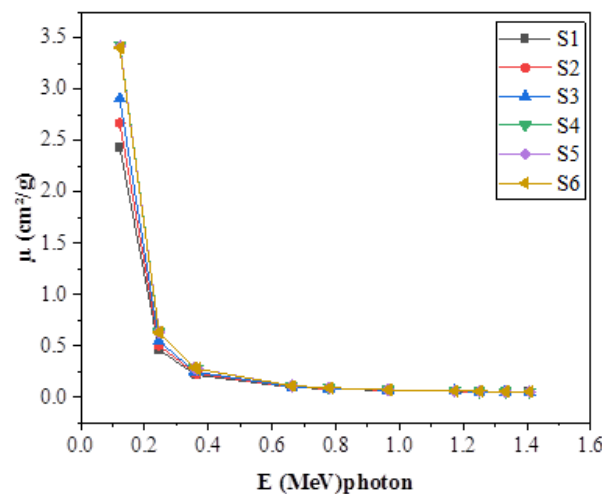


Fig 2: Mass attenuation coefficient of the samples under study as a function of incident photon energies.

Table 3: Evaluate the molecular cross section with incident photon energies.

E(MeV)	σ_m (Barn/molecule)					
	S1	S2	S3	S4	S5	S6
0.05953	1641.594	1678.756	1711.276	1798.131	1789.004	1780.234
0.08099	735.9361	756.4451	774.9351	823.6401	819.4858	815.3373
0.12178	694.0049	800.836	914.6311	1183.335	1178.183	1173.383
0.24469	132.7823	151.6489	171.6939	219.062	218.1157	217.205
0.35601	64.18047	72.16522	80.60836	100.4178	100.0214	99.62554
0.364	61.7844	69.40578	77.43059	96.30997	95.91708	95.55919
0.66166	27.48636	29.93987	32.50134	38.38591	38.24957	38.14786
0.7789	23.35884	25.26683	27.24386	31.75813	31.66885	31.58312
0.96413	19.34542	20.76475	22.22865	25.53077	25.46752	25.40434
1.17323	16.58709	17.71137	18.8621	21.42983	21.38389	21.33455
1.25	15.86827	16.91953	17.99372	20.39079	20.34573	20.30073
1.3325	15.22646	16.21768	17.22917	19.47947	19.43864	19.39786
1.40801	14.73298	15.68379	16.65025	18.79598	18.75919	18.71899
2.75	11.60953	12.36647	13.13897	14.84002	14.81352	14.78359
7.12	11.95183	12.83437	13.74306	15.7617	15.73095	15.70024
10	12.95305	13.94414	14.96698	17.23569	17.20368	17.17171
20	15.93958	17.22247	18.55377	21.51958	21.47701	21.43448
30	17.95342	19.42402	20.9481	24.35365	24.3052	24.2568
40	19.40818	21.0137	22.67543	26.39376	26.34012	26.28653

50	20.51493	22.21945	23.98744	27.94025	27.88528	27.82692
60	21.3992	23.17925	25.02886	29.16224	29.10278	29.04338
70	22.11517	23.96209	25.87522	30.15986	30.0961	30.03584
80	22.71418	24.61296	26.58314	30.98834	30.92386	30.85945
90	23.22477	25.16785	27.18094	31.68909	31.62401	31.559
100	23.66405	25.64475	27.69693	32.28973	32.22414	32.15861
200	26.09435	28.27821	30.54748	35.62433	35.55934	35.45994
300	27.16973	29.44197	31.80915	37.07416	37.00792	36.94174
400	27.78872	30.11383	32.5328	37.93715	37.83568	37.7688
500	28.19662	30.56374	33.00475	38.48947	38.38753	38.32017
600	28.49328	30.86368	33.35084	38.86918	38.80141	38.69923
700	28.72433	31.10363	33.60255	39.17986	39.07733	39.00938
800	28.89548	31.31359	33.79132	39.4215	39.31876	39.2506

Fig 3 shows the changes in molecular cross-sectional area of the studied samples depending on the incident photon energies. The results indicate a direct relationship between the molecular cross-sectional area and the molecular weight fraction of the samples; the higher the molecular weight fraction, the higher the molecular cross-sectional area, and vice versa.

The results showed that sample S4, composed of (60Pb40Bi), had the highest molecular cross-sectional area. This is attributed to the fact that this sample has the highest molecular weight fraction value (207.912) among all the samples. In contrast, sample S1, composed of (60Pb40Sn), recorded the lowest molecular cross-sectional area due to its lower molecular weight fraction (171.804) compared to the other samples.

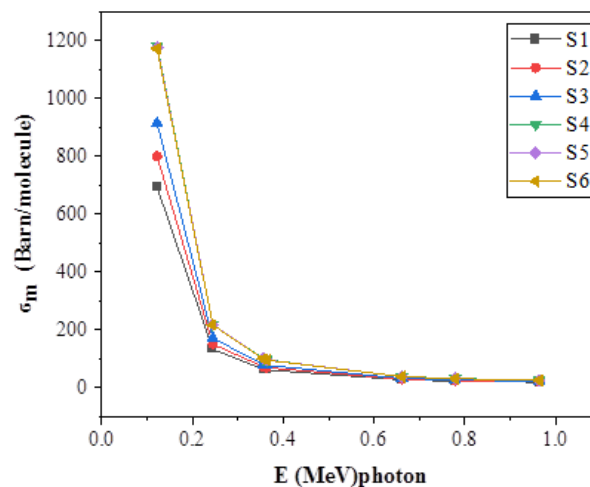


Fig 3: Molecular cross-sectional area of the studied samples as a function of incident photon energies.

Table 4: Evaluate the electron cross section with incident photon energies

E(MeV)	σ_{ele} (Barn/electron)					
	S1	S2	S3	S4	S5	S6
0.05953	23.5606	23.05559	22.54258	21.82221	21.73779	21.65749
0.08099	10.56235	10.38882	10.20819	9.995736	9.95739	9.919015
0.12178	9.960546	10.99847	12.0484	14.36102	14.31584	14.27483
0.24469	1.905727	2.082706	2.261718	2.658547	2.650275	2.642415
0.35601	0.921136	0.991098	1.061851	1.218675	1.215338	1.211998
0.364	0.886747	0.953201	1.01999	1.168822	1.165467	1.162529
0.66166	0.394492	0.411186	0.428139	0.465853	0.464762	0.464089

0.7789	0.335252	0.347008	0.358882	0.385418	0.384801	0.384226
0.96413	0.277651	0.285178	0.292817	0.309843	0.30945	0.309057
1.17323	0.238062	0.243243	0.24847	0.260073	0.259831	0.259546
1.25	0.227746	0.232368	0.237031	0.247464	0.247217	0.246969
1.3325	0.218534	0.222729	0.226959	0.236404	0.236195	0.235985
1.40801	0.211452	0.215397	0.219333	0.228109	0.227939	0.227726
2.75	0.166623	0.169838	0.173079	0.180099	0.179996	0.17985
7.12	0.171536	0.176264	0.181037	0.191285	0.191143	0.191002
10	0.185906	0.191505	0.197159	0.209173	0.209038	0.208903
20	0.228769	0.236529	0.244408	0.261163	0.260962	0.260762
30	0.257672	0.266764	0.275949	0.295557	0.295327	0.295097
40	0.278551	0.288597	0.298703	0.320316	0.320053	0.31979
50	0.294436	0.305156	0.315986	0.339084	0.338828	0.338529
60	0.307127	0.318338	0.329704	0.353914	0.353621	0.353328
70	0.317403	0.329089	0.340853	0.366022	0.365691	0.365402
80	0.326	0.338028	0.350179	0.376076	0.375749	0.375422
90	0.333328	0.345649	0.358054	0.38458	0.384256	0.383932
100	0.339633	0.352198	0.364851	0.39187	0.391548	0.391227
200	0.374513	0.388365	0.402401	0.432339	0.432074	0.431389
300	0.389947	0.404348	0.419021	0.449934	0.449675	0.449416
400	0.398831	0.413575	0.428554	0.460407	0.459733	0.459478
500	0.404686	0.419754	0.43477	0.46711	0.466438	0.466185
600	0.408943	0.423874	0.43933	0.471718	0.471467	0.470797
700	0.412259	0.427169	0.442645	0.475489	0.47482	0.47457
800	0.414716	0.430053	0.445132	0.478421	0.477754	0.477505

Fig 4 illustrates how the electron cross-sectional area of the studied samples changes depending on the incident photon energies. A direct relationship was observed between the electron cross-sectional area and the mass-to-atomic-number ratio; the higher the mass-to-atomic-number ratio, the larger the cross-sectional area, and vice versa.

The data indicated that sample S4, composed of (60Pb40Bi), had the highest electron cross-sectional area, due to its highest mass-to-atomic-number ratio among the other samples. The results showed that sample S1, composed of (60Pb40Sn), had the lowest electron cross-sectional area, due to its lower mass-to-atomic-number ratio compared to the other samples.

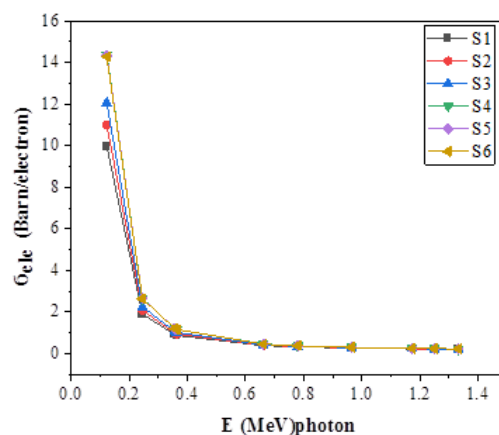


Fig4: Electron cross-sectional area of the samples under study as a function of incident photon energies.

Table 5: Evaluate linear attenuation coefficient with incident photon energies.

E(MeV)	$\mu(\text{cm}^{-1})$					
	S1	S2	S3	S4	S5	S6
0.05953	55.98464	56.70321	57.29443	55.81964	56.39306	56.97065
0.08099	25.09824	25.55038	25.94524	25.56838	25.83187	26.09225
0.12178	23.66822	27.04977	30.62234	36.73445	37.13875	37.55034
0.24469	4.528384	5.122234	5.748404	6.800374	6.875453	6.950948
0.35601	2.1888	2.437519	2.698811	3.117284	3.15288	3.188195
0.364	2.107085	2.344313	2.592417	2.989764	3.023503	3.058064
0.66166	0.93739	1.011276	1.088162	1.191619	1.205705	1.2208
0.7789	0.796626	0.853435	0.912139	0.985872	0.998267	1.010716
0.96413	0.659753	0.701369	0.744227	0.792555	0.802788	0.812984
1.17323	0.565683	0.598236	0.631513	0.665249	0.674064	0.682743
1.25	0.541169	0.57149	0.602439	0.632994	0.641339	0.649659
1.3325	0.519281	0.547783	0.576842	0.604704	0.612746	0.620766
1.40801	0.502451	0.52975	0.557459	0.583486	0.591328	0.599041
2.75	0.39593	0.417701	0.4399	0.460681	0.466952	0.473101
7.12	0.407603	0.433505	0.460125	0.489293	0.495872	0.502436
10	0.441748	0.47099	0.501102	0.53505	0.542295	0.549525
20	0.543601	0.581722	0.62119	0.668035	0.676999	0.685942
30	0.61228	0.656084	0.701354	0.756014	0.76615	0.776261
40	0.661893	0.709778	0.759185	0.819345	0.830295	0.841216
50	0.699638	0.750504	0.803112	0.867353	0.879001	0.890511
60	0.729795	0.782924	0.83798	0.905288	0.917379	0.92944
70	0.754212	0.809366	0.866316	0.936257	0.948691	0.9612
80	0.774641	0.83135	0.890018	0.961975	0.974784	0.987557
90	0.792054	0.850092	0.910032	0.983729	0.996854	1.009944
100	0.807035	0.866201	0.927308	1.002375	1.015771	1.029133
200	0.889917	0.955151	1.022746	1.105891	1.120903	1.134781
300	0.926592	0.994459	1.064987	1.150898	1.166566	1.182202
400	0.947702	1.017152	1.089216	1.177688	1.192658	1.208669
500	0.961613	1.032349	1.105017	1.194834	1.210054	1.226314
600	0.97173	1.04248	1.116604	1.206622	1.2231	1.238444
700	0.97961	1.050585	1.125031	1.216266	1.231798	1.24837
800	0.985446	1.057676	1.131352	1.223767	1.239408	1.256089

Fig 5 shows how the linear attenuation coefficient of the tested samples changes depending on the incident photon energies. The results indicate that the linear attenuation coefficient increases with increasing sample density and decreases with decreasing sample density, indicating a direct relationship between the two.

This observation is supported by the results for individual samples. Sample S6, composed of (80Pb20Bi) and characterized by its highest density (11.028), exhibited the highest linear attenuation coefficient. In contrast, sample S1, composed of (60Pb40Sn) and characterized by its lowest density (9.728), recorded the lowest linear attenuation coefficient among all the samples studied.

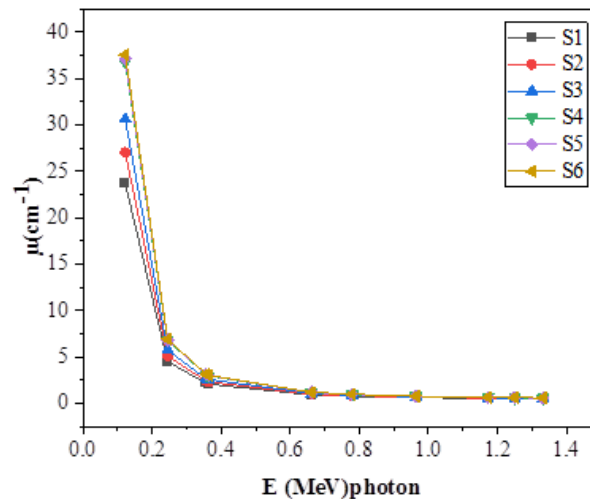


Fig 5: Linear attenuation coefficient of the samples under study as a function of incident photon energies.

Table 6: Effective atomic number values of the samples under study .

Z_{eff}					
S1	S2	S3	S4	S5	S6
69.67538	72.8134	75.91305	82.39914	82.29925	82.19943

Fig 6 shows the behavior of the effective atomic number of the studied samples as a function of changing incident photon energies. It was observed that the effective atomic number is inversely proportional to the electron cross-sectional area and, conversely, directly proportional to the atomic cross-sectional area. This means that increasing the atomic cross-sectional area leads to a corresponding increase in the effective atomic number, and vice versa in the case of a decrease.

The results of the study confirm this relationship, as sample S4 (consisting of 60Pb40Bi) exhibited the highest effective atomic number and the highest atomic cross-sectional area (1183.335) compared to the other samples. Meanwhile, sample S1 (consisting of 60Pb40Sn) recorded the lowest effective atomic number, which corresponds to the lowest atomic cross-sectional area (694.0049) among the samples.

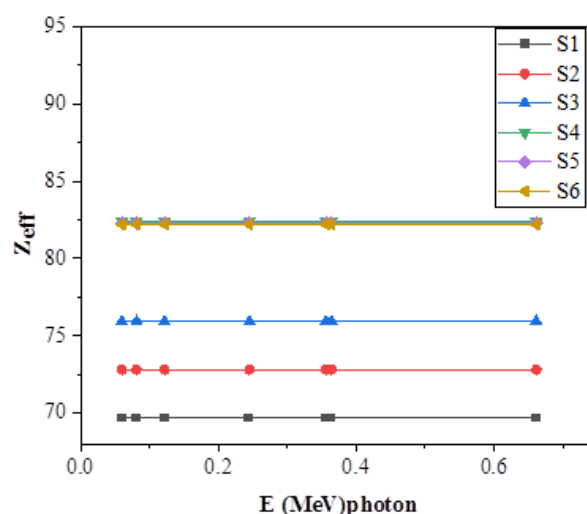


Fig 6: Effective atomic number of the studied samples as a function of incident photon energies.

Table 7: Electron density values of the samples under study .

$N_{ele} * 10^{23}$					
S1	S2	S3	S4	S5	S6
2.4426	2.4276	2.4176	2.367	2.3861	2.3853

Fig 7 shows the variation of electron density in the samples under study as a function of the incident photon energies. The results show a high degree of similarity in electron density values between the samples, indicating that this density does not change with the incident photon energy, as it depends primarily on the effective atomic number.

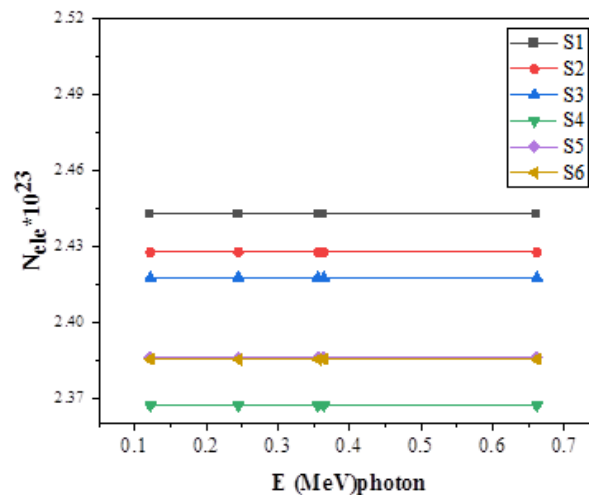


Fig 7: Electron density of the samples under study as a function of the incident photon energies.

Table 8: Evaluate mean free path with incident photon energies.

E(MeV)	$\lambda(\text{cm})$					
	S1	S2	S3	S4	S5	S6
0.05953	0.017862	0.017636	0.017454	0.017915	0.017733	0.017553
0.08099	0.039843	0.039138	0.038543	0.039111	0.038712	0.038326
0.12178	0.042251	0.036969	0.032656	0.027222	0.026926	0.026631
0.24469	0.220829	0.195227	0.173961	0.147051	0.145445	0.143865
0.35601	0.456871	0.410253	0.370534	0.320792	0.31717	0.313657
0.364	0.474589	0.426564	0.38574	0.334475	0.330742	0.327004
0.66166	1.066792	0.988849	0.918981	0.839194	0.82939	0.819135
0.7789	1.255294	1.171735	1.096324	1.01433	1.001736	0.989397
0.96413	1.515719	1.425783	1.343676	1.261742	1.245658	1.230036
1.17323	1.767774	1.671582	1.583498	1.503196	1.483539	1.464679
1.25	1.847853	1.749813	1.659918	1.579794	1.559237	1.539268
1.3325	1.925741	1.82554	1.733577	1.653702	1.631998	1.610913
1.40801	1.990243	1.887683	1.793853	1.713837	1.691109	1.669335
2.75	2.525702	2.394056	2.273245	2.1707	2.141546	2.113713
7.12	2.453366	2.306776	2.173322	2.043767	2.01665	1.990305
10	2.263732	2.123186	1.9956	1.868985	1.844014	1.819753
20	1.839586	1.719034	1.609813	1.496927	1.477106	1.45785
30	1.633239	1.524196	1.425814	1.322727	1.305228	1.288227
40	1.510818	1.408892	1.317201	1.220487	1.204392	1.188756
50	1.429311	1.332437	1.245156	1.152933	1.137655	1.122951
60	1.370249	1.277264	1.193346	1.104621	1.090062	1.075917

70	1.325887	1.235536	1.154313	1.068083	1.054084	1.040366
80	1.290921	1.202863	1.123573	1.039528	1.025869	1.012599
90	1.262541	1.176343	1.098862	1.01654	1.003156	0.990154
100	1.239104	1.154467	1.07839	0.997631	0.984474	0.971692
200	1.1237	1.046955	0.97776	0.904248	0.892138	0.881227
300	1.079224	1.005572	0.938978	0.868886	0.857217	0.845879
400	1.055184	0.983137	0.918092	0.849121	0.838463	0.827357
500	1.03992	0.968665	0.904964	0.836936	0.82641	0.815452
600	1.029093	0.959251	0.895573	0.82876	0.817595	0.807465
700	1.020815	0.951851	0.888864	0.822189	0.811822	0.801045
800	1.014769	0.945469	0.883899	0.817149	0.806837	0.796122

Fig 8 illustrates the dependence of the mean free path of the studied samples on the incident photon energies. The results indicate an inverse relationship between the mean free path and the linear attenuation coefficient; the mean free path value decreases with increasing attenuation coefficient. Sample S1, composed of (60Pb40Sn), was found to have the highest mean free path due to its lower linear attenuation coefficient compared to the other samples. In contrast, sample S6, composed of (80Pb20Bi), recorded the lowest mean free path due to its highest linear attenuation coefficient among the samples.

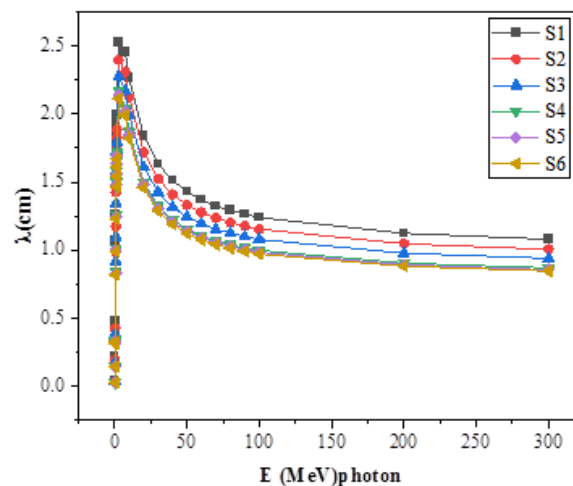


Fig 8: Mean free path of the samples under study as a function of the incident photon energies.

Table 9: Evaluate half-value layer with incident photon energies.

E(MeV)	HVL(cm)					
	S1	S2	S3	S4	S5	S6
0.05953	0.012381	0.012224	0.012098	0.012418	0.012291	0.012167
0.08099	0.027617	0.027129	0.026716	0.02711	0.026833	0.026565
0.12178	0.029286	0.025625	0.022635	0.018869	0.018664	0.018459
0.24469	0.153067	0.135321	0.120581	0.101928	0.100815	0.09972
0.35601	0.316679	0.284366	0.256834	0.222356	0.219846	0.21741
0.364	0.32896	0.295672	0.267375	0.23184	0.229253	0.226662
0.66166	0.739443	0.685418	0.636989	0.581685	0.574889	0.567781
0.7789	0.870103	0.812184	0.759914	0.70308	0.69435	0.685798
0.96413	1.050616	0.988277	0.931365	0.874573	0.863424	0.852596
1.17323	1.225327	1.158652	1.097597	1.041936	1.02831	1.015238
1.25	1.280834	1.212877	1.150567	1.095029	1.080781	1.066939

1.3325	1.334822	1.265367	1.201624	1.146259	1.131214	1.116599
1.40801	1.379531	1.308442	1.243404	1.187941	1.172187	1.157094
2.75	1.750682	1.659433	1.575693	1.504615	1.484406	1.465114
7.12	1.700544	1.598935	1.506432	1.416631	1.397835	1.379574
10	1.569099	1.47168	1.383244	1.295481	1.278172	1.261356
20	1.275103	1.191543	1.115837	1.03759	1.023852	1.010504
30	1.132075	1.056492	0.988299	0.916845	0.904715	0.89293
40	1.047219	0.976569	0.913014	0.845977	0.834821	0.823982
50	0.990723	0.923575	0.863076	0.799152	0.788562	0.77837
60	0.949784	0.885332	0.827164	0.765665	0.755573	0.745769
70	0.919035	0.856408	0.800109	0.740339	0.730635	0.721126
80	0.894798	0.833761	0.778801	0.720546	0.711078	0.70188
90	0.875126	0.815379	0.761673	0.704612	0.695335	0.686322
100	0.858881	0.800215	0.747483	0.691505	0.682385	0.673525
200	0.778889	0.725694	0.677731	0.626777	0.618383	0.61082
300	0.748061	0.697009	0.65085	0.602266	0.594177	0.586319
400	0.731398	0.681458	0.636373	0.588566	0.581178	0.57348
500	0.720817	0.671427	0.627273	0.58012	0.572823	0.565228
600	0.713312	0.664902	0.620763	0.574453	0.566713	0.559692
700	0.707575	0.659773	0.616114	0.569898	0.562712	0.555242
800	0.703384	0.655349	0.612672	0.566404	0.559257	0.551829

Fig 9 represents the behavior of the half-thickness layer for the studied samples as a function of incident photon energies. The results show that the half-thickness layer has an inverse relationship with the linear attenuation coefficient; a higher linear attenuation coefficient leads to a lower value for the half-thickness layer, and vice versa. Sample S1, with a composition of (60Pb40Sn), was found to have the highest value for the half-thickness layer, attributed to its lower linear attenuation coefficient compared to the other samples. In contrast, sample S6, composed of (80Pb20Bi), recorded the lowest value for the half-thickness layer due to its highest linear attenuation coefficient among all studied samples.

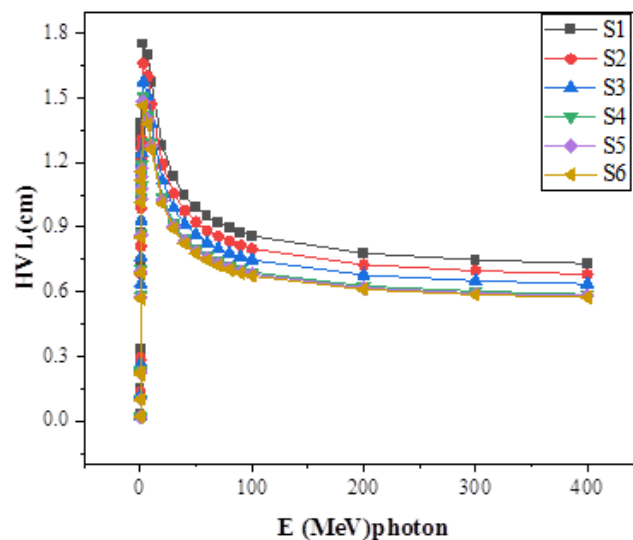


Fig 9: Half-Value layer for the studied samples as a function of incident photon energies.

Table 10: Evaluate tenth layer thickness with incident photon energies.

E(MeV)	TVL(cm)
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	S1	S2	S3	S4	S5	S6
0.05953	0.041129	0.040608	0.040189	0.04125	0.040831	0.040417
0.08099	0.091743	0.090119	0.088748	0.090056	0.089137	0.088248
0.12178	0.097286	0.085124	0.075193	0.062682	0.062	0.06132
0.24469	0.508478	0.449528	0.400561	0.338597	0.334899	0.331262
0.35601	1.051985	0.944643	0.853185	0.738651	0.730312	0.722222
0.364	1.092782	0.9822	0.8882	0.770156	0.761562	0.752955
0.66166	2.456379	2.27691	2.116031	1.932316	1.909742	1.886129
0.7789	2.890422	2.698019	2.524379	2.335582	2.306582	2.278172
0.96413	3.490071	3.282986	3.093928	2.905267	2.868234	2.832263
1.17323	4.07045	3.84896	3.646139	3.461237	3.415974	3.372548
1.25	4.254838	4.029093	3.822102	3.637609	3.590276	3.544295
1.3325	4.434182	4.203461	3.991709	3.80779	3.757814	3.709263
1.40801	4.582704	4.34655	4.130499	3.946255	3.893921	3.843786
2.75	5.815642	5.512518	5.234339	4.998222	4.931091	4.867003
7.12	5.649085	5.311547	5.004258	4.705947	4.643508	4.582845
10	5.212434	4.888817	4.595039	4.303496	4.245998	4.190135
20	4.235803	3.958222	3.706732	3.446801	3.401162	3.356824
30	3.760671	3.509591	3.283058	3.045692	3.005398	2.966251
40	3.478787	3.244092	3.03968	2.810274	2.773214	2.737211
50	3.29111	3.06805	2.867078	2.654726	2.619547	2.58569
60	3.155114	2.941008	2.747781	2.543484	2.509959	2.47739
70	3.052969	2.844926	2.657904	2.459352	2.427119	2.39553
80	2.972456	2.769694	2.587123	2.393601	2.36215	2.331596
90	2.907107	2.70863	2.530223	2.340671	2.309853	2.279913
100	2.853142	2.658259	2.483085	2.29713	2.266835	2.237403
200	2.587414	2.410703	2.251375	2.082108	2.054223	2.0291
300	2.485004	2.315415	2.162077	2.000685	1.973815	1.947709
400	2.429651	2.263756	2.113985	1.955173	1.930632	1.905059
500	2.394503	2.230433	2.083756	1.927117	1.902879	1.877648
600	2.369573	2.208757	2.062132	1.908291	1.882581	1.859256
700	2.350513	2.191718	2.046685	1.893159	1.869288	1.844474
800	2.336591	2.177022	2.035251	1.881555	1.85781	1.833138

Figure 10 shows the behavior of the tenth layer thickness (Tenth Value Layer, or TVL) for the samples under study as a function of incident photon energies. The results revealed a clear inverse relationship between the tenth layer thickness and the linear attenuation coefficient; the higher the linear attenuation coefficient, the lower the tenth layer thickness, and vice versa. This is because the linear attenuation coefficient represents the ability of a material to reduce radiation intensity through absorption or scattering. Therefore, materials with high attenuation coefficients require a smaller thickness to achieve a significant reduction in photon intensity. Sample S1, composed of (60Pb40Sn), showed the highest value for the tenth layer thickness due to its lower linear attenuation coefficient compared to the other samples. Sample S6, composed of (80Pb20Bi), recorded the lowest value for the tenth layer thickness due to its high linear attenuation coefficient. These results are of great practical importance in the design of radiation shields, as the tenth-layer thickness is used as a benchmark for determining the thickness of shielding materials required to reduce radiation intensity to one-tenth of its original value. Therefore, choosing materials with high linear attenuation coefficients allows for reducing the required shield thickness, contributing to improved efficiency and reducing cost and space.

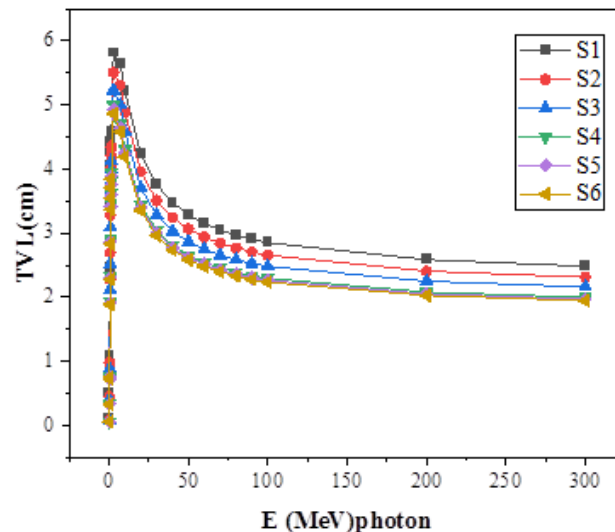


Fig 10: Tenth layer thickness for the studied samples as a function of incident photon energies.

4. Conclusion

By evaluating a set of gamma ray protection parameters, including mass attenuation coefficient, molecular and electron cross-sectional area, linear attenuation coefficient, effective atomic number, electron density, mean free path, and layer thickness for Pb-Bi and Pb-Sn alloys with varying component ratios, the following conclusions were reached:

- Most of the gamma-ray shielding properties of the studied samples are closely related to the mass attenuation coefficient, which in turn depends on the incident photon energy, showing high values at low energies and then decreasing significantly with increasing energy.
- At high energies around $E > 100 \text{ MeV}$, most shielding parameters exhibit attenuation stability.
- The elements that make up the studied alloys exhibit distinct absorption edges as follows:
 1. Lead (Pb): The M-type absorption edge lies between 0.002484 and 0.003851 MeV, the L-type edge lies between 0.01304 and 0.01586 MeV, and the K-type absorption edge lies at 0.088 MeV.
 2. Bismuth (Bi): The M-type absorption edge lies between 0.00258 and 0.003999 MeV, the L-type edge lies between 0.01304 and 0.01639 MeV, and the K-type edge lies at 0.09053 MeV.
 3. Tin (Sn): The L-type absorption edge lies between 0.003929 and 0.004465 MeV, and the K-type edge lies at 0.2920 MeV.
- Sample S1, with composition (60Pb40Sn), recorded the highest values for mean free path (0.042251), half layer thickness (0.029286), and tenth layer thickness (0.097286), while it recorded the lowest value for linear attenuation coefficient (23.66822).
- Sample S6, with composition (80Pb20Bi), exhibited excellent attenuation properties, recording the lowest mean free path (0.026631), half layer thickness (0.018459), and tenth layer thickness (0.06132). Concurrently, this sample had the highest mass attenuation coefficient (3.405). Together, these results indicate that sample S6 is the ideal alloy choice for gamma-ray shielding applications, due to its superior radiation attenuation capability.

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